Enantioselective synthesis of allenic esters via an ylide route†

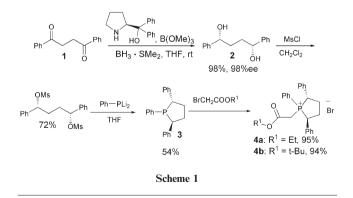
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Pseudo- C_2 -symmetric chiral phosphorus ylides have been designed and synthesized for the enantioselective preparation of allenic esters, and up to 92% ee has been achieved.

Chiral ylides have developed into excellent reagents for the synthesis of optically active epoxides,¹ cyclopropanes,² aziridines³ and olefins.⁴ Although the Wittig or Horner-Wadsworth-Emmons (HWE) reaction is good for the preparation of allenes,⁵ which are versatile building blocks⁶, as well as the basic units of numerous natural products and biologically-active compounds,⁷ only a few examples involve its asymmetric version.^{5c-g} Recently, Tanaka et al. described that optically-active 4,4-disubstituted allenecarboxylates could be prepared with binol-derived HWE reagents in 21-71% yields with 23-89% ee values. 5e-5f Pinho e Melo and his co-workers^{5g} documented that a phosphorus ylide bearing a 10-phenylsulfonyll-iso-borneol reacted with methylketene to give a penta-2.3-dienoic ester with excellent diastereoselectivity. In our studies of ylides in organic synthesis,⁸ we are interested in developing the chiral ylide-mediated syntheses of optically active allenes9 and have found that newly-designed chiral phosphorus ylides 7a and 7b could react with ketene very well to afford allenes with ee values up to 92%. In this communication, we wish to report the preliminary results.

Phosphonium salts **4a** and **4b** were readily available from the corresponding diketone on a gram scale (Scheme 1). The diketone **1** was reduced with a proline-derived chiral borane to give diol **2** in 98% yield with 98% ee.¹⁰ The diol was treated with methanesulfonyl chloride,¹⁰ followed by reaction with dilithium phenylphosphate to afford chiral phosphine **3**, which reacted with bromoacetate to give the desired phosphonium salts **4a** and **4b**. Gratifyingly, after deprotonation by NaHMDS, salt **4a** reacted



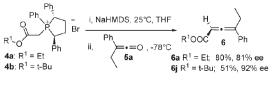
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smoothly with ethyl phenyl ketene at -78 °C to afford chiral allene 6a with 81% ee in 80% yield. Further studies showed that the enantioselectivity could be improved to 92% ee when ethyl ester was replaced by tert-butyl ester (Scheme 2). This result encouraged us to study the generality of the reaction by investigating a variety of ketenes with different structures. As shown in Table 1, the substituents on the ketene proved to strongly influence both the vields and the enantioselectivities. For example, 2-aryl-1-buten-1one gave 4-aryl-2,3-hexadienoic esters in high yields with high enantioselectivities (Table 1, entries 1 and 6-7), and up to 91% ee was obtained. Replacement of the ethyl group on ketene 5a with other alkyl groups, such as iso-propyl, benzyl and allyl, decreased the enantiomeric excess greatly (Table 1, entries 2 and 4-5). As dialkyl ketene 5h and monosubstituted ketene 5i are not stable and could not be purified, as could ketenes 5a-5c, we developed a onepot strategy,¹¹ finding that in the presence of triethylamine, the reaction of acid chloride with phosphonium salt 4a worked well, giving the desired allenic esters in good yields with moderate enantioselectivities (Table 1, entries 8-9).

The absolute configurations of **6a** and **6e** were assigned as *S* by chemical transformations.¹¹ Very recently, Aggarwal and his coworkers developed an elegant dipole–dipole interaction model to account for the selectivity of the reaction of stabilized phosphorus ylides with aldehydes.¹² Based on their mechanistic insights, as well as the current experimental results, a stereochemical model, shown in Scheme 3, has been developed to explain the configuration of the allenic esters formed. This reaction was proposed to proceed *via* a $_{\pi}2_{s} + _{\pi}2_{a}$ cycloaddition, affording the oxaphosphetane, which then decomposed into the corresponding phosphine oxide and chiral allene. The ketene approached the *re* face of ylide **7a** preferentially due to the steric hindrance between the substituents of the ketene and the ylide. Thus allenes with *S*-configurations were obtained, consistent with the experimental results.¹¹

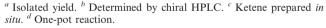
One of the advantages of the current reaction is that the chiral phosphine oxide could be recycled and reused. For example, **8** was recovered in 81% yield and was readily reduced by LiAlH₄ into phosphine **3** in 90% yield (Scheme 4).¹¹

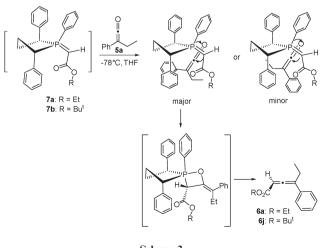
In summary, we have developed an efficient method for the preparation of optically-active 4,4-disubstituted and 4-monosubstituted allenic esters in good yields. The easily available phosphonium salts, good to high enantioselectivity and, in particular, the recovery and the reuse of the phosphine oxide



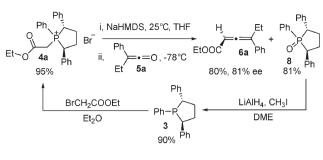
4					
	O Ph Ph P Br	i, 1	NaHMDS, 25°C, THF	Н <u>_</u>	R ¹
R-	4a : R = Et 4b : R = t-Bu	ii,	$R^1 \rightarrow = 0$, -78°C $R^2 5$	800C 6	R ²
Entry	Ketene	4	6	Yield (%) ^a	ee $(\%)^b$
1	o=∙≠ ^{Et}	4a	H Et	80	81
	5a ^{Ph}		EtO ₂ C 6a Ph		
2	O=∙⇒⊂	4 a	H_ Pr ⁱ	76	71
	5b Ph		EtO ₂ C 6b Ph		
3	O=∙ = ∕	4 a	H Bu ⁱ	71	85
	5c ^{`Ph}		EtO ₂ C 6c Ph		
4 ^{<i>c</i>}	0=•	4 a		46	61
5 ^c	5d ^{Ph} Bn	4a	EtO ₂ C ⁶ d ^{Ph} H Bn	51	52
Ū	0=•=			01	02
6	5e ^{Ph} Et	4a	EtO ₂ C <mark>6e</mark> Ph HEt	78	91
	0=•= 5f		EtO ₂ C 6f		
7	Et	4a	H Et	75	85
	0=•=				
8^d	5g CI		- S CI	51	(2)
8"	O=∙ 	4 a	HEt	51	63
9^d	5h ^{`Bun} H	4a	EtO ₂ C [°] 6h [`] Bu ⁿ Н Н	52	55
フ	0=•=<		´,=•=<	52	55
10	5i C ₁₀ H ₂₁ r	4b	$EtO_2C_{6i} C_{10}H_{21}^n$ H. Et	51	92
10	0=• € t	40		51	92
	5a ^{Ph}		Bu'O ₂ C 6j Ph		

Table 1Asymmetric synthesis of allenes 6 from phosphonium salts 4^{11}









Scheme 4

make the present method potentially useful. Studies on further improving the enantioselectivity and understanding the mechanism are in progress in our laboratory.

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Notes and references

- For reviews, see: (a) A.-H. Li, L.-X. Dai and V. K. Aggarwal, Chem. Rev., 1997, 97, 2641; (b) L.-X. Dai, X.-L. Hou and Y.-G. Zhou, Pure Appl. Chem., 1999, 71, 369; (c) V. K. Aggarwal and C. L. Winn, Acc. Chem. Res., 2004, 37, 611. Selected examples, see: (d) V. K. Aggarwal, E. Alonso, I. Bae, G. Hynd, K. M. Lydon, M. J. Palmer, M. Patel, M. Porcelloni, J. Richardson, R. A. Stenson, J. R. Studley, J.-L. Vasse and C. L. Winn, J. Am. Chem. Soc., 2003, 125, 10926; (e) K. Li, X.-M. Deng and Y. Tang, Chem. Commun., 2003, 2704; (f) M. Davoust, J.-F. Brière, P.-A. Jaffrès and P. Metzner, J. Org. Chem., 2005, 70, 4166; (g) V. K. Aggarwal, J. P. H. Charmant, D. Fuentes, J. N. Harvey, G. Hynd, D. Ohara, W. Picoul, R. Robiette, C. Smith, J.-L. Vassa and C. L. Winn, J. Am. Chem. Soc., 2006, 128, 2105.
- 2 For reviews, see: (a) Y.-Z. Huang, Y. Tang and Z.-L. Zhou, *Tetrahedron*, 1998, **53**, 1667; (b) H. Lebel, J. F. Marcoux, C. Molinaro and A. B. Charette, *Chem. Rev.*, 2003, **103**, 977; (c) S. Ye, Y. Tang and X.-L. Sun, *Synlett*, 2005, 2720. For recent examples, see: (d) C. D. Papageorgious, S. V. Ley and M. J. Gaunt, *Angew. Chem., Int. Ed.*, 2003, **42**, 828; (e) N. Bremeyer, S. C. Smith, S. V. Ley and M. J. Gaunt, *Angew. Chem., Int. Ed.*, 2004, **43**, 2681; (f) C. D. Papageorgious, M. A. Cubillo de Dios, S. V. Ley and M. J. Gaunt, *Angew. Chem., Int. Ed.*, 2004, **43**, 4641; (g) R. K. Kunz and D. W. C. Macmillan, *J. Am. Chem. Soc.*, 2005, **127**, 3240; (h) K. Huang and Z.-Z. Huang, *Synlett*, 2005, 1621; (i) J.-C. Zheng,

W.-W. Liao, Y. Tang, X.-L. Sun and L.-X. Dai, J. Am. Chem. Soc., 2005, 127, 12222.

- 3 For recent examples, see: (a) X.-F. Yang, M.-J. Zhang, X.-L. Hou and L.-X. Dai, J. Org. Chem., 2002, 67, 8097; (b) V. K. Aggarwal and J.-L. Vasse, Org. Lett., 2003, 5, 3987; (c) D. Morton, D. Pearson, R. A. Field and R. A. Stockman, Org. Lett., 2004, 6, 2377; (d) L. G. Arini, A. Sinclair, P. Szeto and R. A. Stockman, Tetrahedron Lett., 2004, 45, 1589; (e) J.-C. Zheng, W.-W. Liao, X.-X. Sun, X.-L. Sun, Y. Tang, L.-X. Dai and J.-G. Deng, Org. Lett., 2005, 7, 5789.
- 4 (a) J. Aubé, Chemtracts: Org. Chem., 1988, 1, 461; (b) M. Yamaguchi and M. Hirama, Chemtracts: Org. Chem., 1994, 7, 401; (c) T. Rein and O. Reiser, Acta Chem., Scand., 1996, 50, 369; (d) W.-M. Dai, A. Wu and H. Wu, Tetrahedron: Asymmetry, 2002, 13, 2187.
- 5 (a) H. J. Bestmann and H. Hartung, Chem. Ber., 1966, 99, 1198; (b) R. W. Lang and H. J. Hansen, Helv. Chim. Acta, 1980, 63, 438; (c) I. Tömösközi and H. J. Bestmann, Tetrahedron Lett., 1964, 1293; (d) H. J. Bestmann and I. Tömösközi, Tetrahedron, 1968, 24, 3299; (e) K. Tanaka and K. Fuji, Tetrahedron Lett., 1996, 37, 3735; (f) J. Yamazaki, T. Watanabe and K. Tanaka, Tetrahedron: Asymmetry, 2001, 12, 669; (g) T. M. V. D. Pinho e Melo, A. L. Cardoso, A. M. d'A. Rocha Gonsalves, J. C. Pessoa, J. A. Paixão and A. M. Beja, Eur. J. Org. Chem., 2004, 4830.
- 6 For recent reviews, see: (a) N. Krause and A. Hoffmann-Roder, in Modern Organocopper Chemistry, ed. N. Krause, Wiley-VCH, Weinheim, 2002, pp. pp. 145–166; (b) S. Ma, Acc. Chem. Res., 2003, 36, 701; (c) S. Ma, Chem. Rev., 2005, 105, 2829; (d) M. A. Tius,

Cyclizations of Allenes, in Modern Allene Chemistry, ed. N. Krause and A. S. K. Hashmi, Wiley-VCH, Weinheim, 2004, vol. 4, pp. pp. 817-845.

- 7 (a) M. T. Crimmins and K. A. Emmitte, J. Am. Chem. Soc., 2001, 123, 1533; (b) N. Furuichi, H. Hara, T. Osaki, H. Mori and S. Katsumura, Angew. Chem., Int. Ed., 2002, 41, 1023; (c) A. H. Roder and N. Krause, Angew. Chem., Int. Ed., 2004, 43, 1196.
- 8 (a) S. Ye, Y. Tang and L.-X. Dai, J. Org. Chem., 2000, 65, 6257; (b) S. Ye, Y. Tang and L.-X. Dai, J. Org. Chem., 2001, 66, 5717; (c) S. Ye, Z.-Z. Huang, C.-A. Xia, Y. Tang and L. X. Dai, J. Am. Chem. Soc., 2002, 124, 2432; (d) Z.-Z. Huang, S. Ye, W. Xia, Y.-H. Yu and Y. Tang, J. Org. Chem., 2002, 67, 3096; (e) Z.-Z. Huang and Y. Tang, J. Org. Chem., 2002, 67, 5320; (f) W.-W. Liao, K. Li and Y. Tang, J. Am. Chem. Soc., 2003, 125, 13030; (g) W.-W. Liao, X.-M. Deng and Y. Tang, Chem. Commun., 2004, 1516; (h) H. Jiang, X.-M. Deng, X.-L. Sun, Y. Tang and L.-X. Dai, J. Org. Chem., 2005, 70, 10202.
- For recent examples of the synthesis of optically-active allenes, see: (a) P. H. Dixneuf, T. Guyot, M. D. Ness and S. M. Roberts, Chem. Commun., 1997, 2083; (b) Z. K. Sweeney, J. L. Salsman, R. A. Andersen

and R. G. Bergman, Angew. Chem., Int. Ed., 2000, 39, 2339; (c) Z. Wan and S. G. Nelson, J. Am. Chem. Soc., 2000, 122, 10470; (d) N. Krause and M. Purpura, Angew. Chem., Int. Ed., 2000, 39, 4355; (e) M. Ogasawara, H. Ikeda, T. Nagano and T. Hayashi, J. Am. Chem. Soc., 2001, 123, 2089; (f) J. W. Han, N. Tokunaga and T. Hayashi, J. Am. Chem. Soc., 2001, 123, 12915; (g) M. Ogasawara, K. Ueyama, T. Nagano, Y. Mizuhata and T. Hayashi, Org. Lett., 2003, 5, 217; (h) T. Hayashi, N. Tokunaga and K. Inoue, Org. Lett., 2004, 6, 305; (i) B. D. Sherry and F. D. Toste, J. Am. Chem. Soc., 2004, 126, 15978.

- 10 D. J. Aldous, W. M. Dutton and P. G. Steel, Tetrahedron: Asymmetry, 2000, 11, 2455. Compound 2 could also be obtained from ketone 1 on a 10 g scale by asymmetric hydrogenation using Ding's catalyst: Q. Jing, X. Zhang, J. Sun and K. Ding, Adv. Synth. Catal., 2005, 347, 1193.
- For detailed procedures, see the ESI⁺.
 (a) R. Robiette, J. Richardson, V. K. Aggarwal and J. N. Harvey, J. Am. Chem. Soc., 2005, **127**, 13468; (b) R. Robiette, J. Richardson, V. K. Aggarwal and J. N. Harvey, J. Am. Chem. Soc., 2006. 128. 2394.

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